

Sedimentary Rocks



The Prologue

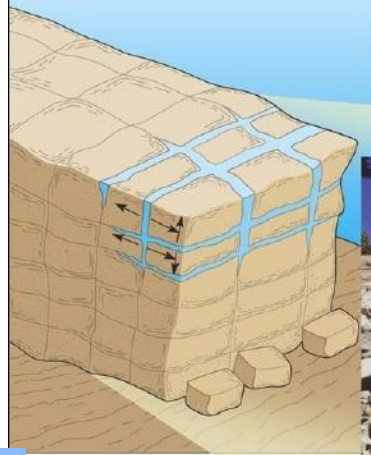
Weathering



Mechanical (particularly important in very cold or dry climate)



Chemical



(a)



(b)



The Chemistry of Weathering

| Most important processes | Examples | Principal kinds of rock materials affected |
|---|--|--|
| Simple (congruent) Solution —Dissolution of soluble minerals in H ₂ O (direct solution) or in H ₂ O + CO ₂ (carbonation) to yield cations and anions in solution | $\text{SiO}_2 + 2\text{H}_2\text{O} \rightarrow \text{H}_4\text{SiO}_4 \text{ (direct solution)}$ (quartz) (silicic acid) aq | Highly soluble minerals (e.g., gypsum, halite), quartz |
| | $\text{CaCO}_3 + \text{H}_2\text{O} + \text{CO}_2 \leftrightarrow \text{Ca}^{2+} + 2\text{HCO}_3^- \text{ (Carbonation)}$ (calcite) aq aq | Carbonate rocks |
| Hydrolysis (incongruent dissolution) —Reaction between H ⁺ and OH ⁻ ions of water and the ions of silicate minerals, yielding soluble cations, silicic acid, and clay minerals (if Al present) | $2\text{KAlSi}_3\text{O}_8 + 2\text{H}^+ + 9\text{H}_2\text{O} \rightarrow \text{H}_4\text{Al}_2\text{Si}_2\text{O}_9 + 4\text{H}_4\text{SiO}_4 + 2\text{K}^+$ (orthoclase) aq (kaolinite) (silicic acid) aq | Silicate minerals |
| | $2\text{NaAlSi}_3\text{O}_8 + 2\text{H}^+ + 9\text{H}_2\text{O} \rightarrow \text{H}_4\text{Al}_2\text{Si}_2\text{O}_9 + 4\text{H}_4\text{SiO}_4 + 2\text{Na}^+$ (albite) aq (kaolinite) (silicic acid) aq | |
| Oxidation —Loss of an electron from an element (commonly Fe or Mn) in a mineral, resulting in the formation of oxides or hydroxides (if water present) | $2\text{FeS}_2 + 15/2\text{O}_2 + 4\text{H}_2\text{O} \rightarrow \text{Fe}_2\text{O}_3 + 4\text{SO}_4^{2-} + 8\text{H}^+$ (pyrite) (hematite) aq aq | Iron- and manganese-bearing silicate minerals, iron sulfides |
| | $\text{MnSiO}_3 + 1/2\text{O}_2 + 2\text{H}_2\text{O} \rightarrow \text{MnO}_2 + \text{H}_4\text{SiO}_4$ (rhodonite) (pyrolusite) (silicic acid) | |
| Other Processes | | |
| Hydration and Dehydration —Gain (hydration) or loss (dehydration) of water molecules from a mineral, resulting in formation of a new mineral | $\text{Fe}_2\text{O}_3 + \text{H}_2\text{O} \leftrightarrow 2\text{FeOOH} \text{ (hydration)}$ (hematite) (goethite) | Ferric oxides |
| | $\text{CaSO}_4 \cdot 2\text{H}_2\text{O} \leftrightarrow \text{CaSO}_4 + 2\text{H}_2\text{O} \text{ (dehydration)}$ (gypsum) (anhydrite) | Evaporites |
| Ion Exchange —Exchange of ions, principally cations, between solutions and minerals | $\text{K-clay} + \text{Mg}^{2+} \leftrightarrow \text{Mg-clay} + \text{K}^+$ $\text{Ca-zeolite} + \text{Na}^+ \leftrightarrow \text{Na-zeolite} + \text{Ca}^{2+}$ | Clay minerals and zeolites |
| Chelation —Bonding of metal ions to organic molecules having ring structures | Metal ions (cations) + chelating agent (e.g., secreted by lichens) → H ⁺ ions + chelate (metal ions/organic molecules in solution) | Silicate minerals |

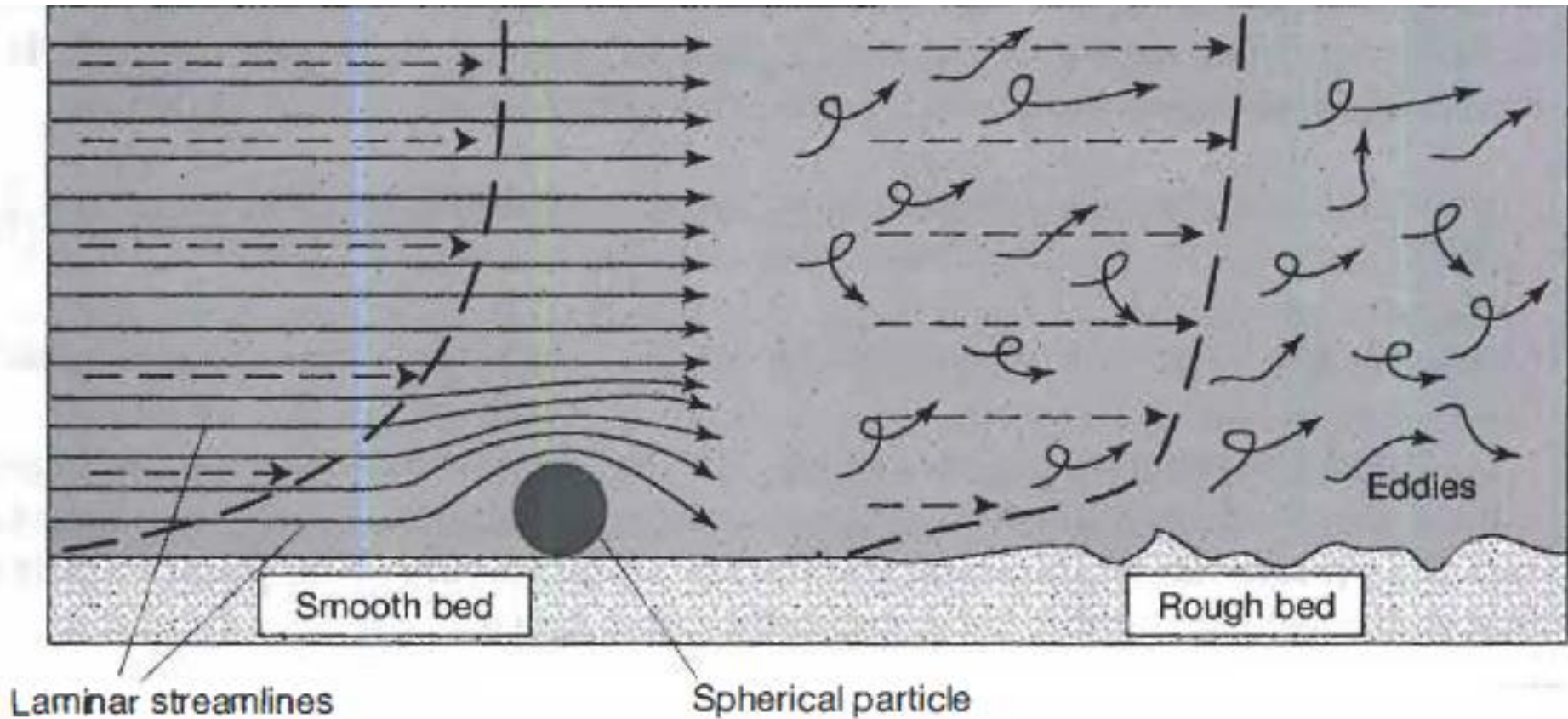
Submarine Weathering Processes and Products

Alteration of ocean rocks occurs both at low temperature (<20°C) as well as at high temperature (~350°C)

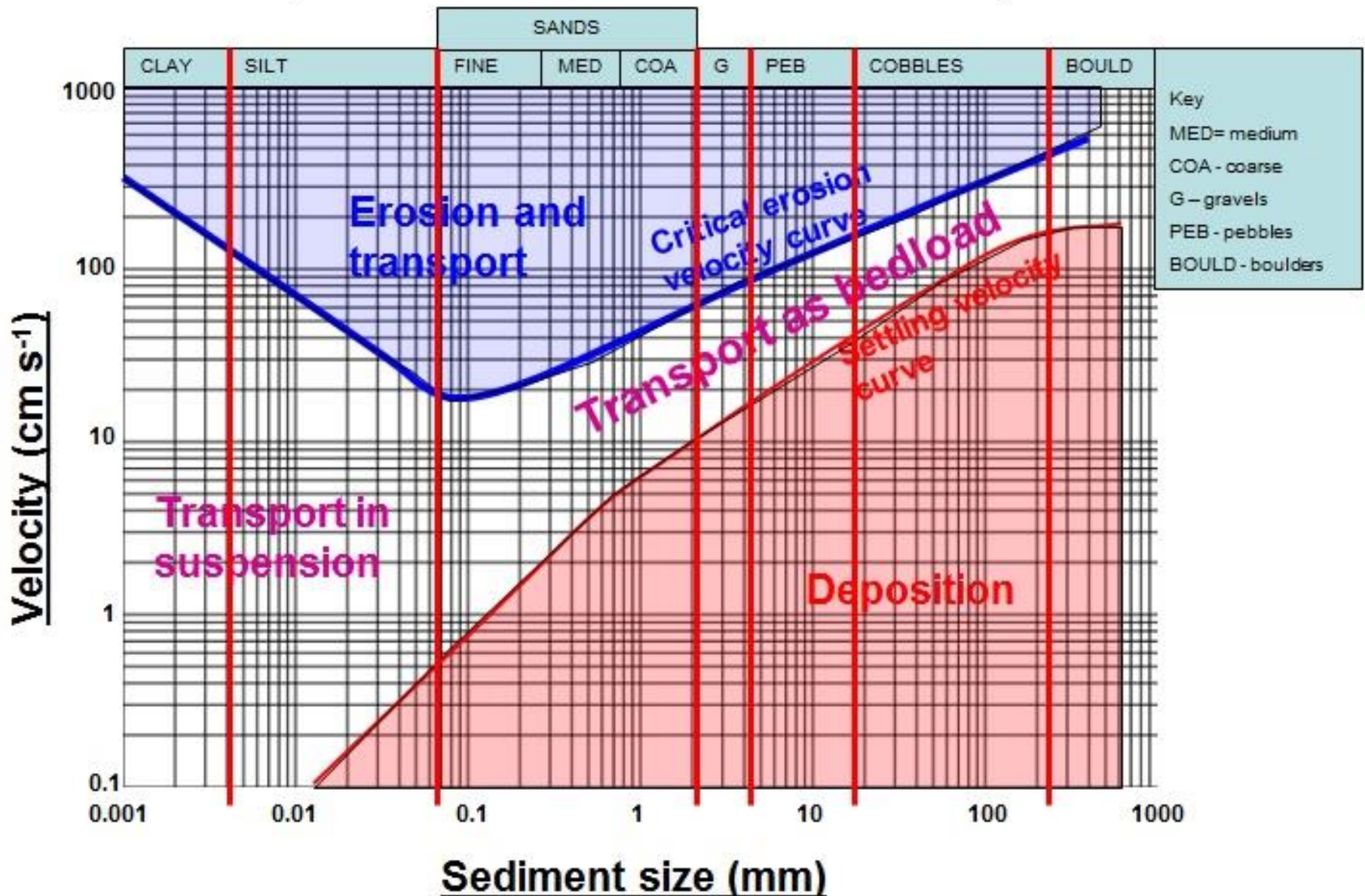
As a result of submarine weathering, chemical elements are exchanged between rock and seawater and large mass of seawater becomes fixed in the oceanic crust in the form of altered hydrous minerals



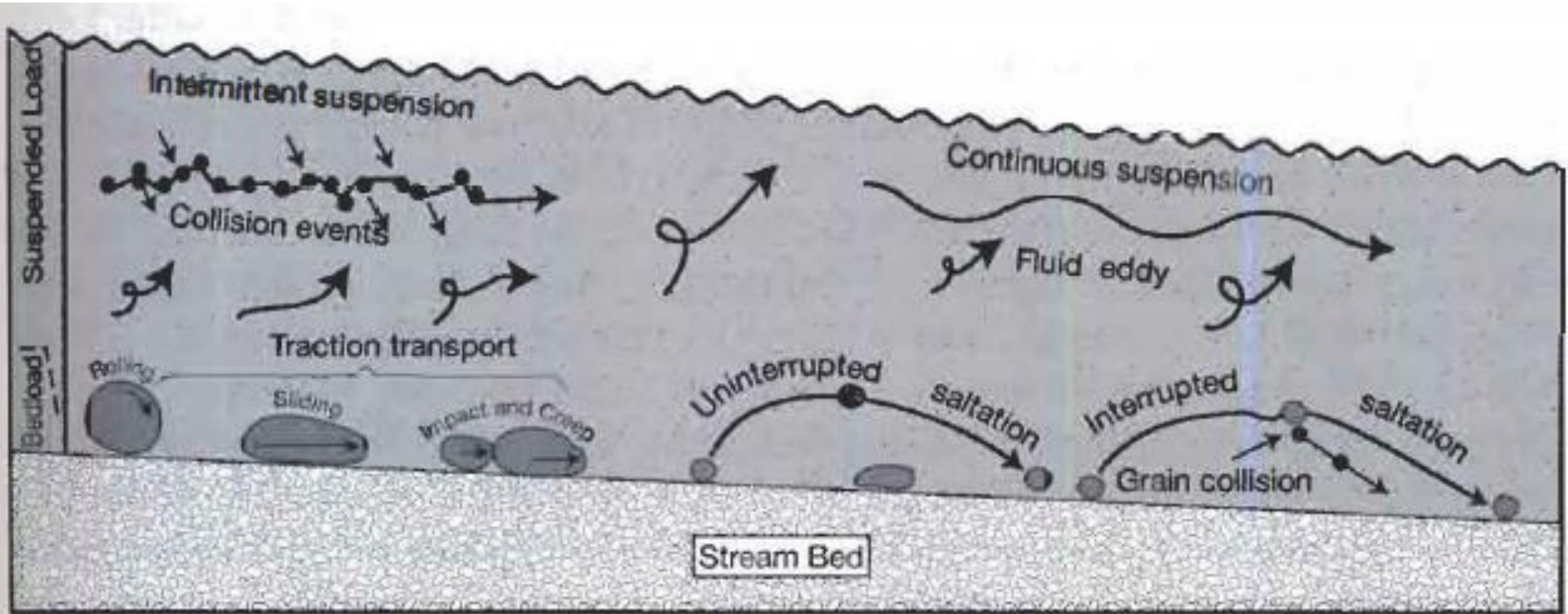
Fluid Flow- Laminar vs. Turbulent



The Hjulström curve



Sediment Transport Path



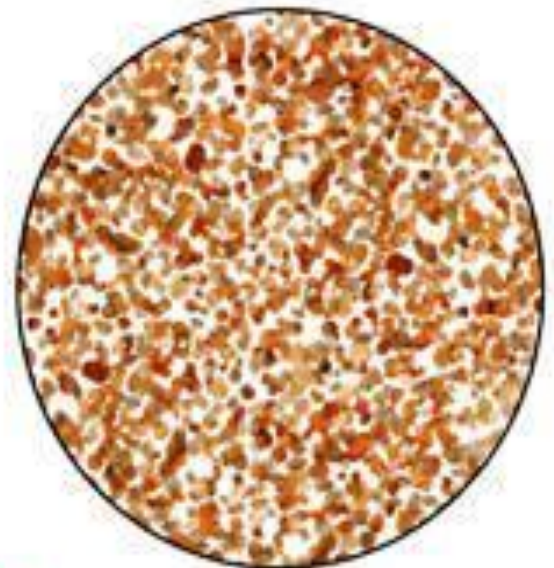
| | U.S. standard sieve mesh | Millimeters | Phi (ϕ) units | Wentworth size class |
|--------|-----------------------------|-------------|-------------------------|----------------------|
| GRAVEL | | 4096 | -12 | |
| | | 1024 | -10 | Boulder |
| | | 256 | -8 | |
| | | 64 | -6 | Cobble |
| | | 16 | -4 | |
| | 5 | 4 | -2 | |
| | 6 | 3.36 | -1.75 | |
| | 7 | 2.83 | -1.5 | Granule |
| | 8 | 2.38 | -1.25 | |
| | 10 | 2.00 | -1.0 | |
| SAND | 12 | 1.68 | -0.75 | |
| | 14 | 1.41 | -0.5 | Very coarse sand |
| | 16 | 1.19 | -0.25 | |
| | 18 | 1.00 | 0.0 | |
| | 20 | 0.84 | 0.25 | |
| | 25 | 0.71 | 0.5 | Coarse sand |
| | 30 | 0.59 | 0.75 | |
| | 35 | 0.50 | 1.0 | |
| | 40 | 0.42 | 1.25 | |
| | 45 | 0.35 | 1.5 | Medium sand |
| | 50 | 0.30 | 1.75 | |
| | 60 | 0.25 | 2.0 | |
| | 70 | 0.210 | 2.25 | |
| | 80 | 0.177 | 2.5 | Fine sand |
| | 100 | 0.149 | 2.75 | |
| | 120 | 0.125 | 3.0 | |
| | 140 | 0.105 | 3.25 | |
| | 170 | 0.088 | 3.5 | Very fine sand |
| | 200 | 0.074 | 3.75 | |
| | 230 | 0.0625 | 4.0 | |
| MUD | 270 | 0.053 | 4.25 | |
| | 325 | 0.044 | 4.5 | Coarse silt |
| | | 0.037 | 4.75 | |
| | | 0.031 | 5.0 | |
| | | 0.0156 | 6.0 | Medium silt |
| | | 0.0078 | 7.0 | Fine silt |
| | | 0.0039 | 8.0 | Very fine silt |
| | | 0.0020 | 9.0 | |
| | | 0.00098 | 10.0 | Clay |
| | | 0.00049 | 11.0 | |
| | 0.00024 | 12.0 | | |
| | 0.00012 | 13.0 | | |
| | 0.00006 | 14.0 | | |

Texture- Grain Size

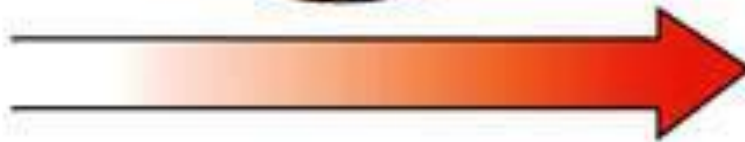
Grain size



Closer to source

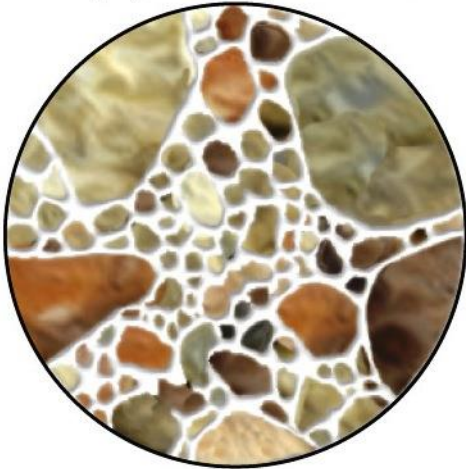


Farther from source

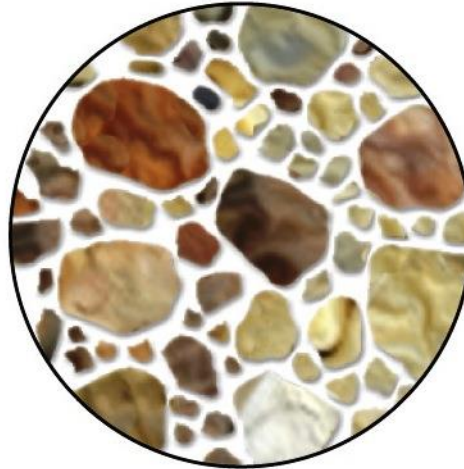


Texture- Grain Sorting

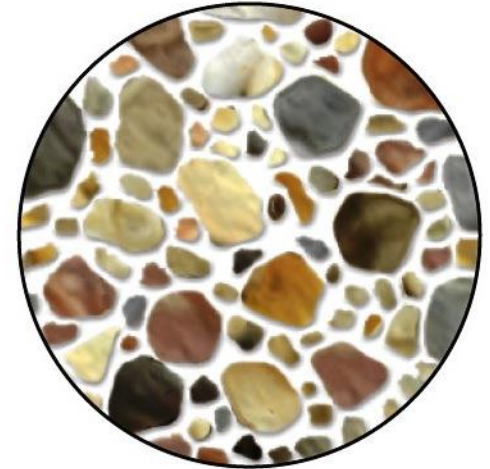
Very poorly sorted



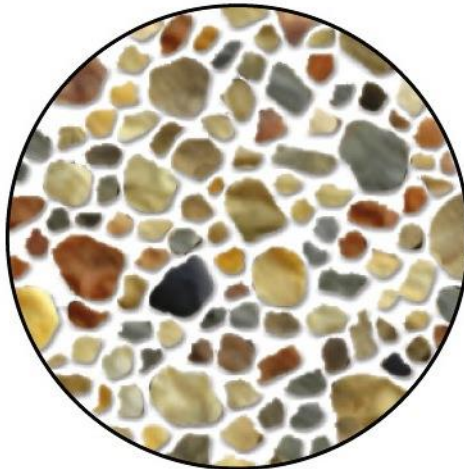
Poorly sorted



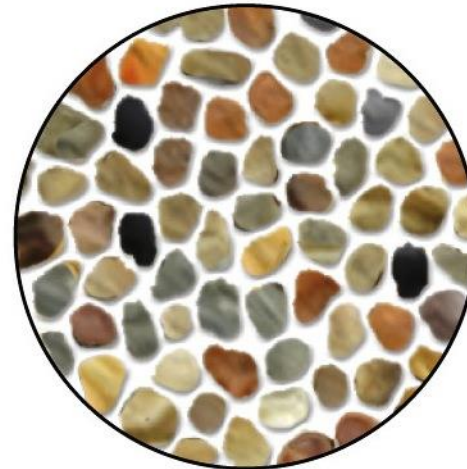
Moderately sorted



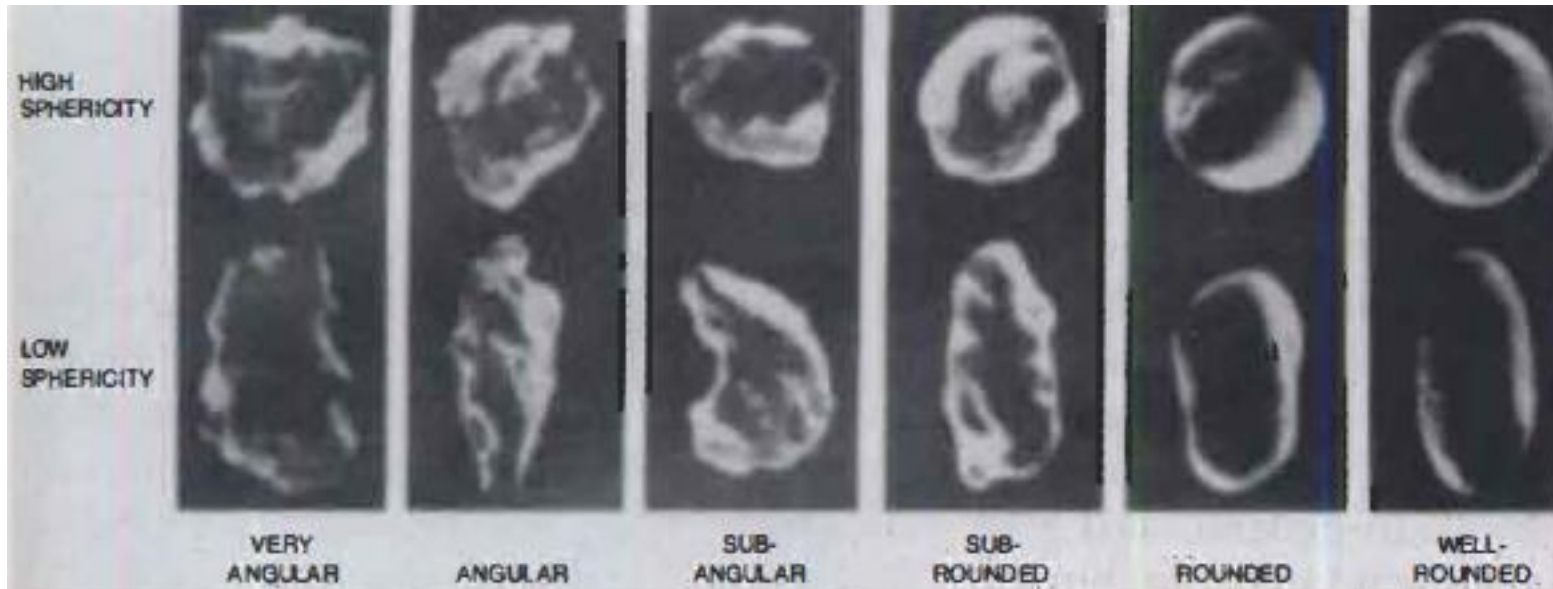
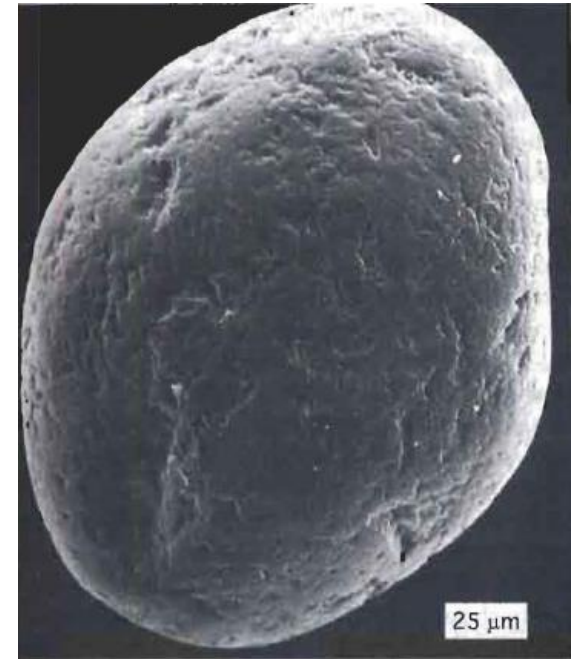
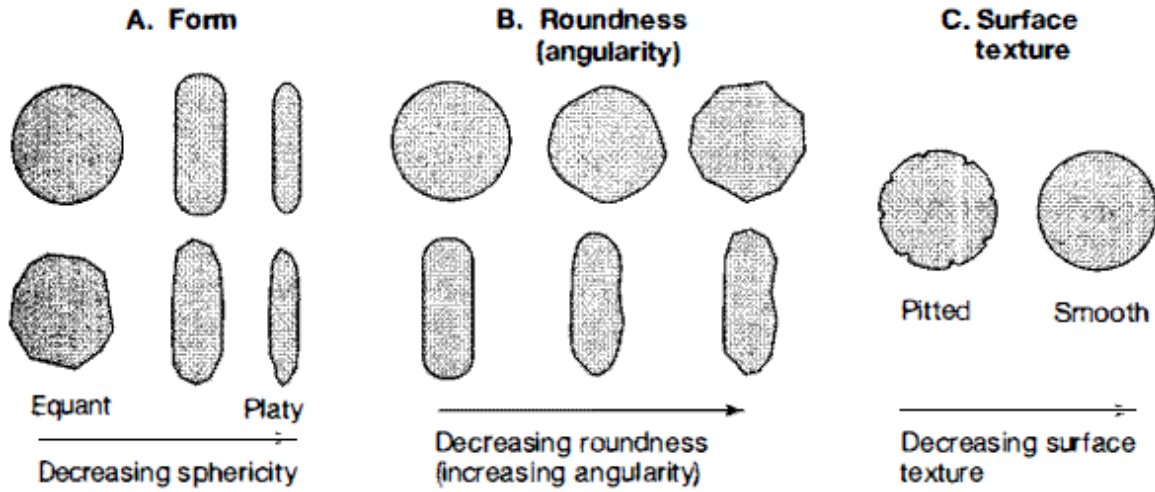
Well sorted



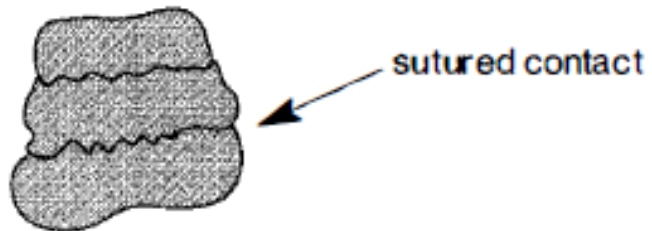
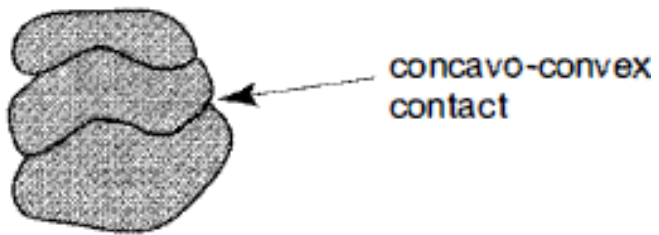
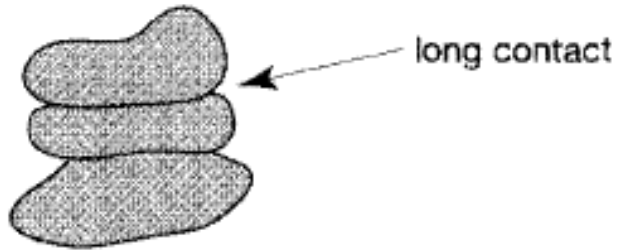
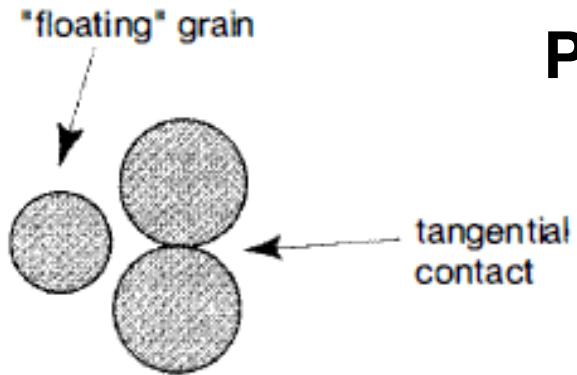
Very well sorted



Texture- Roundness and Sphericity



Principal kinds of Grain Contacts

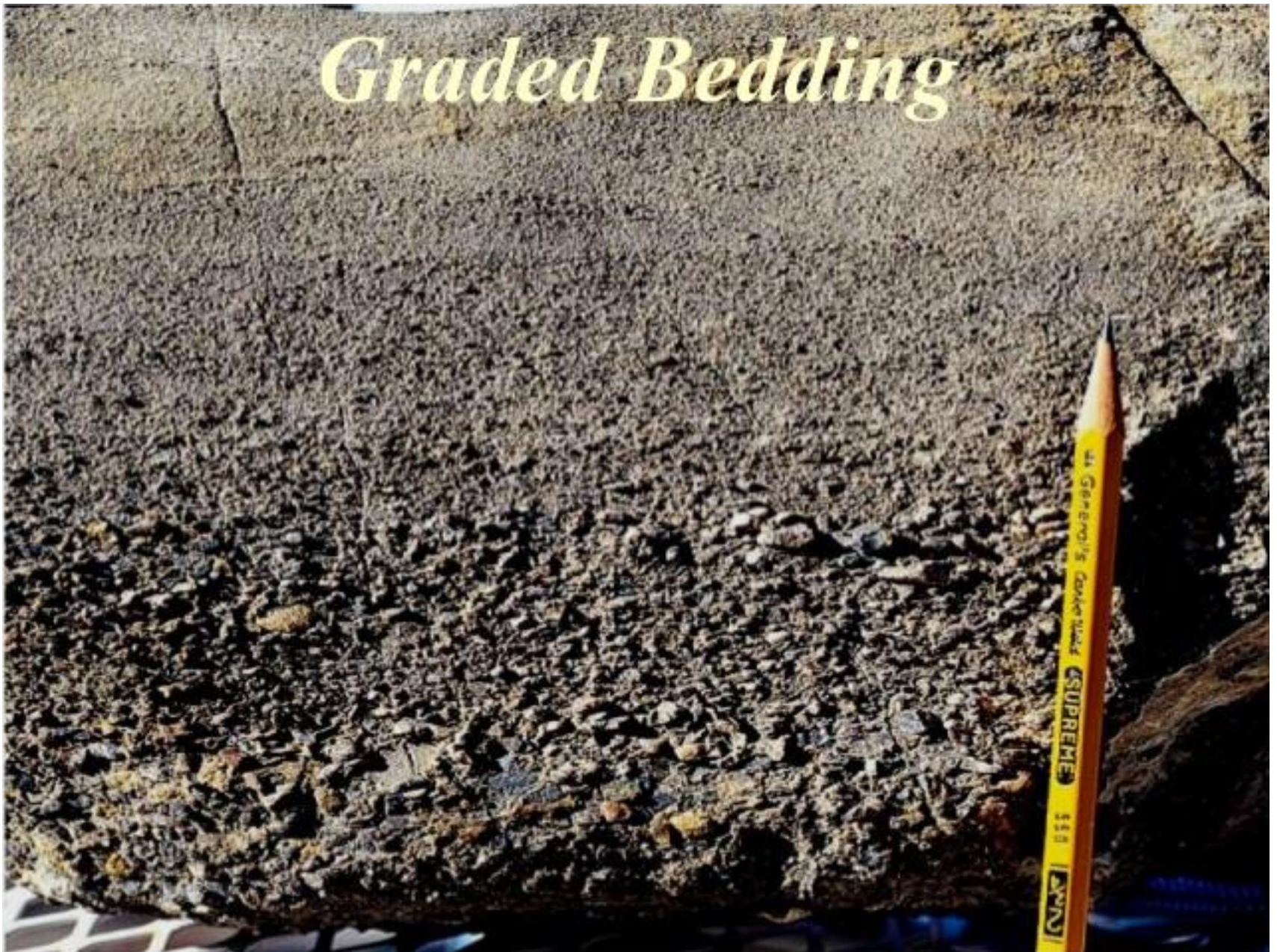


Sedimentary Structures- Bedding



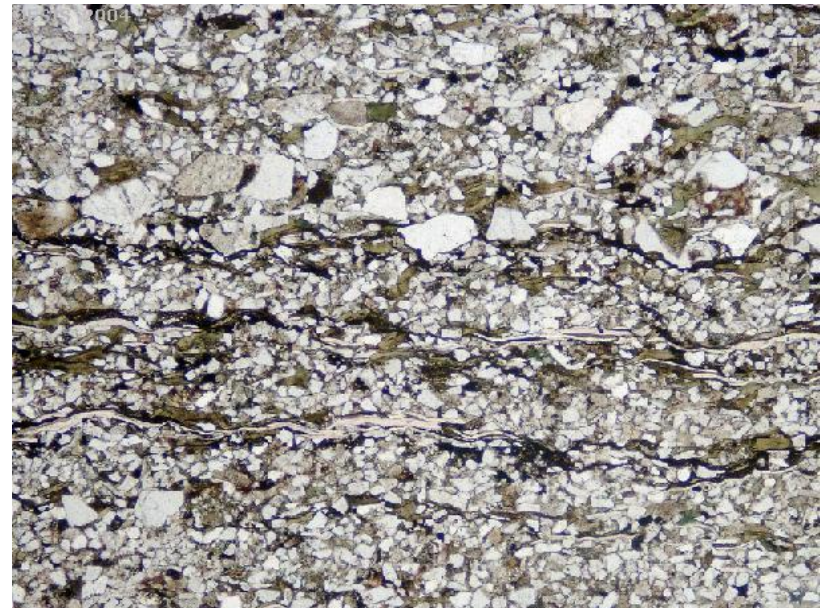
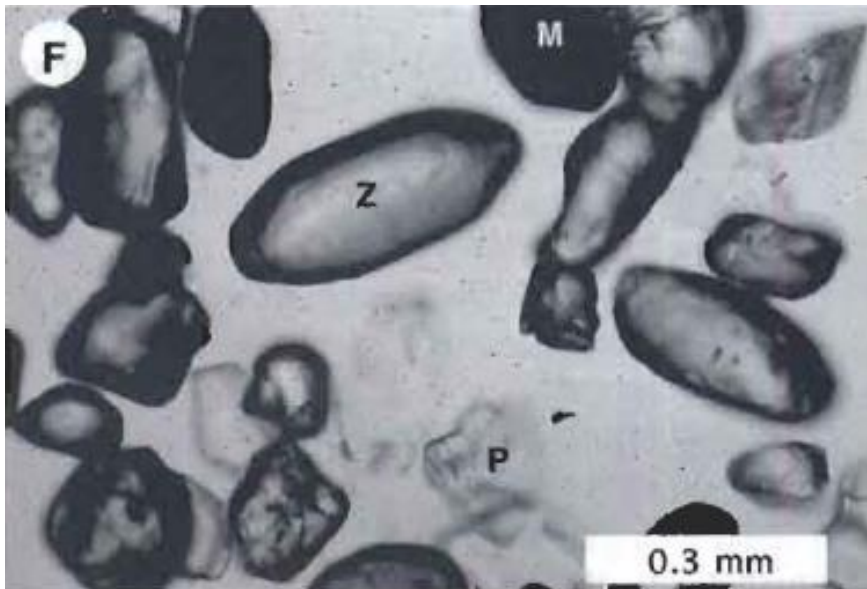
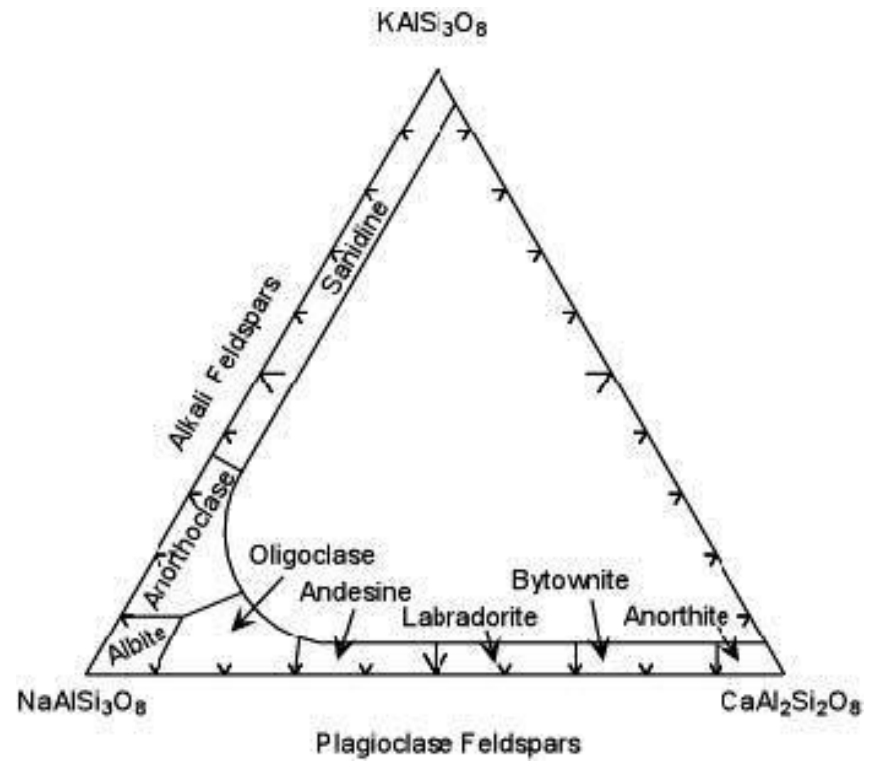
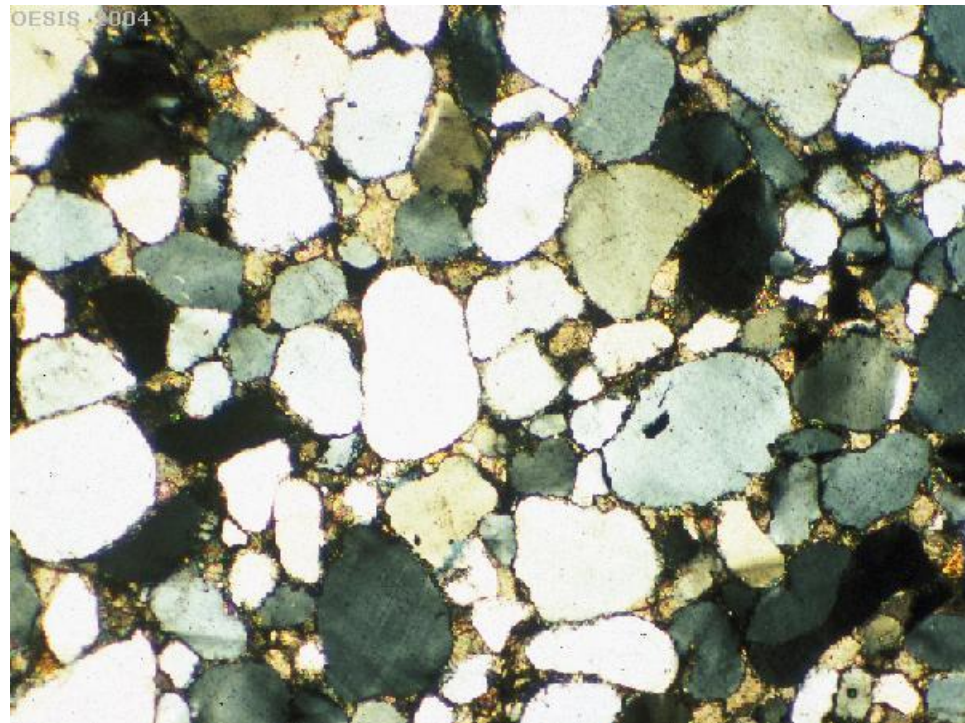
Laminated Bedding

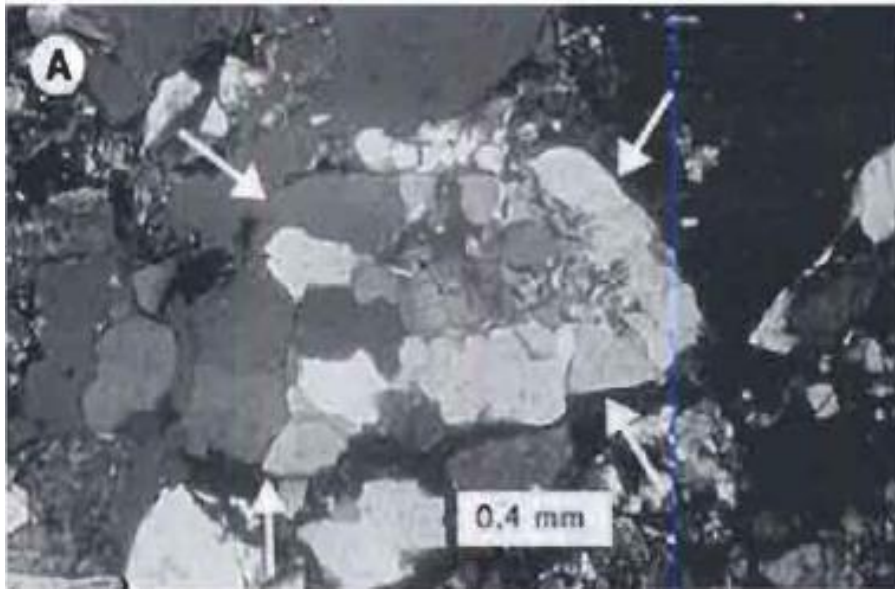
Graded Bedding



Massive Bedding



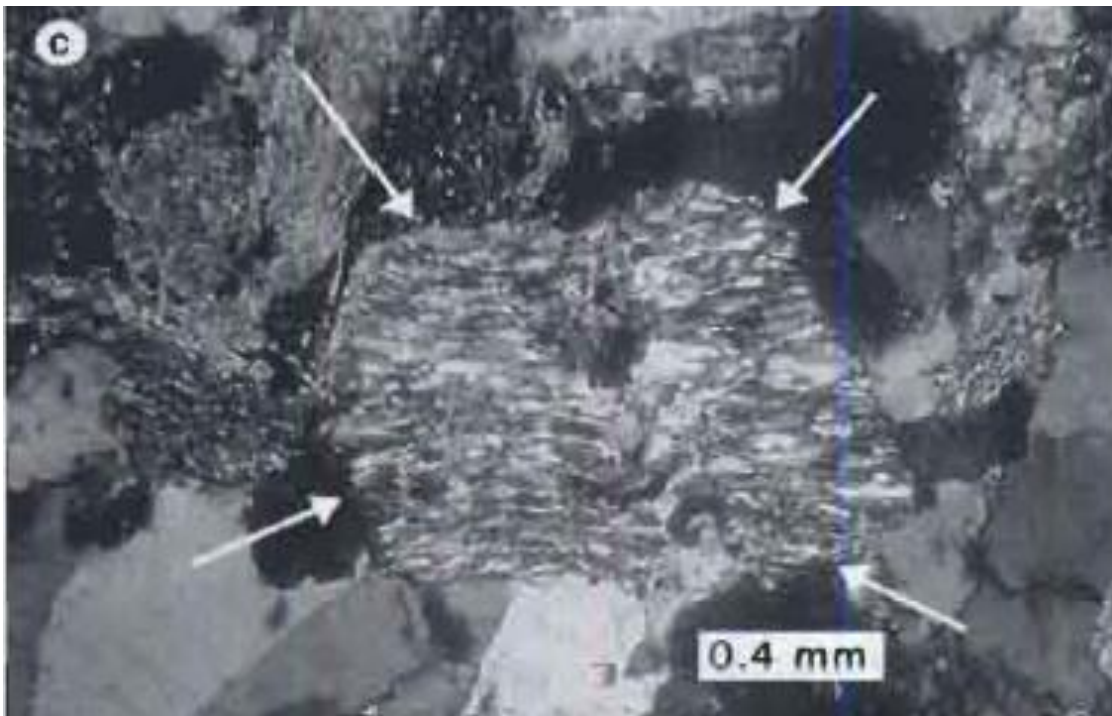




Plutonic

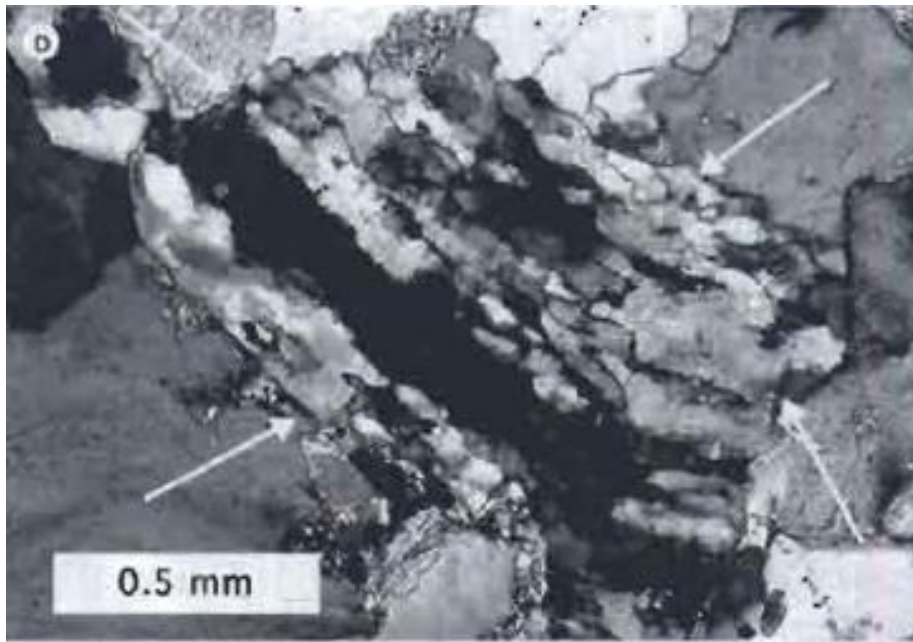


Volcanic

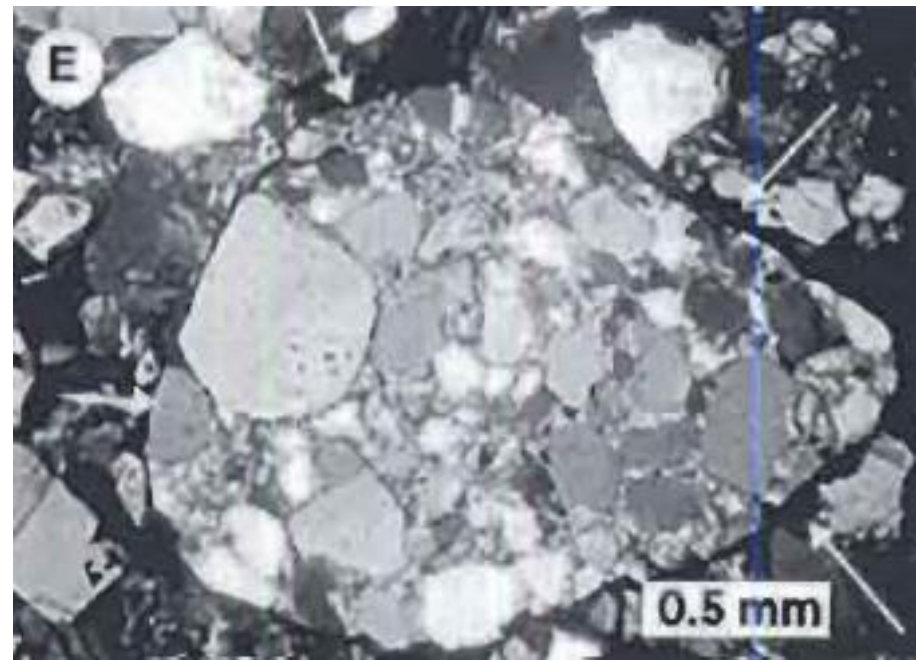


Lithic Fragments

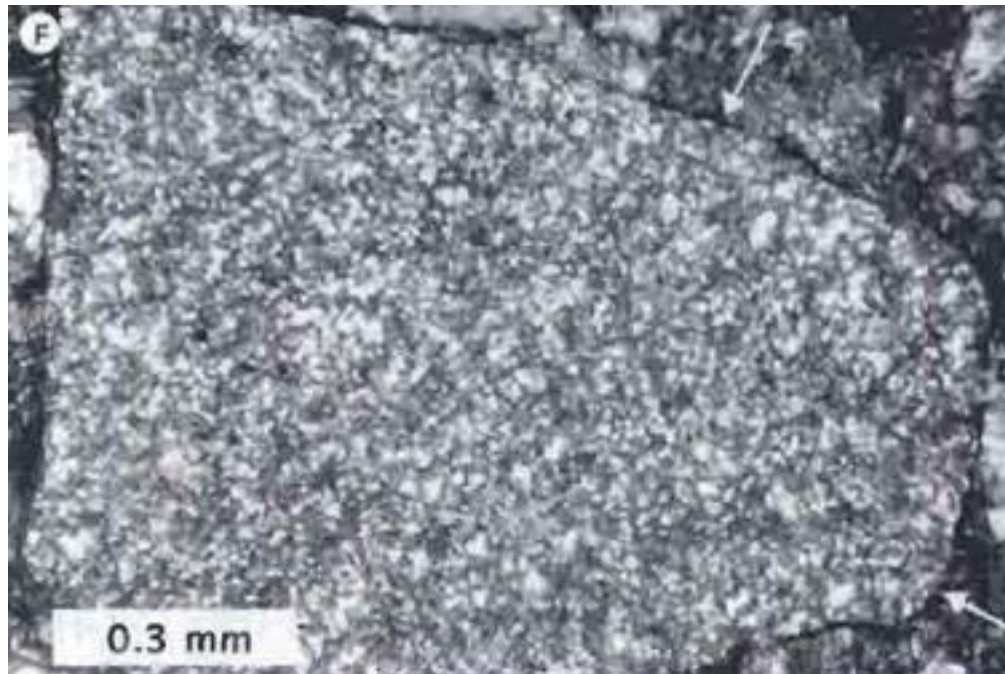
Metamorphic schist



Metamorphic Quartzite



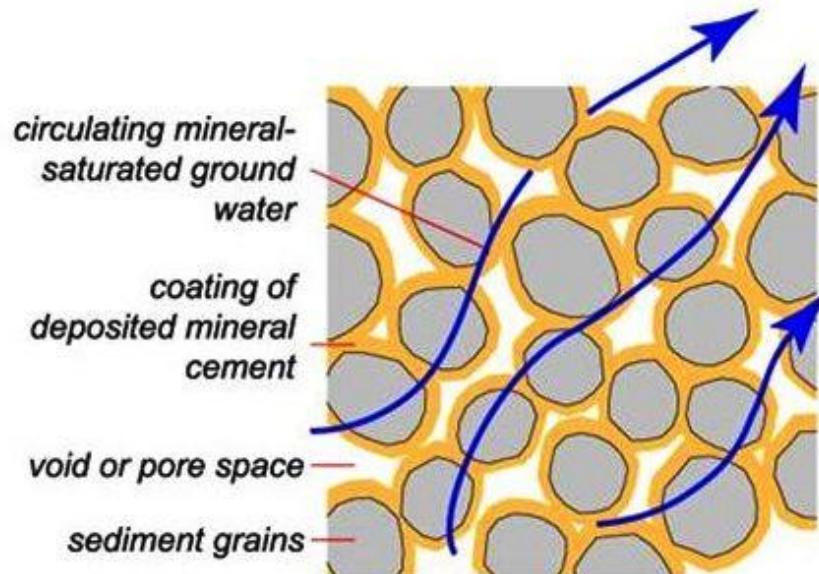
Sandstone



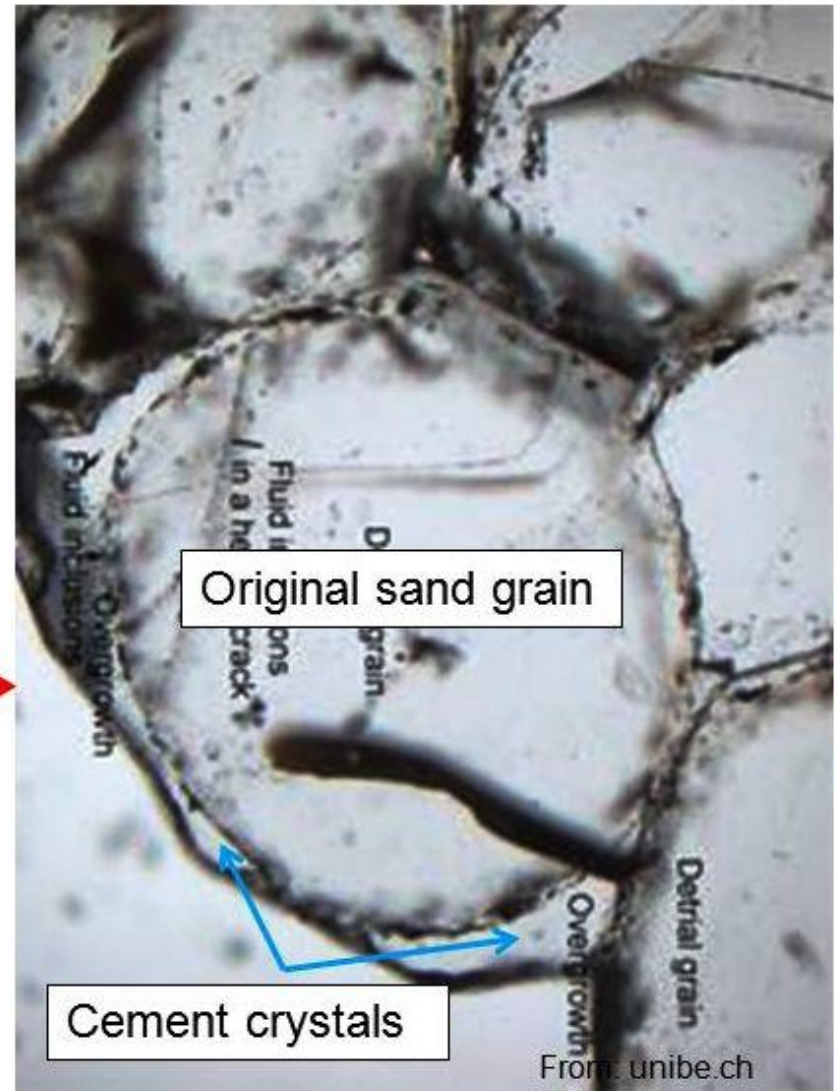
**Lithic
Fragments**

Chert

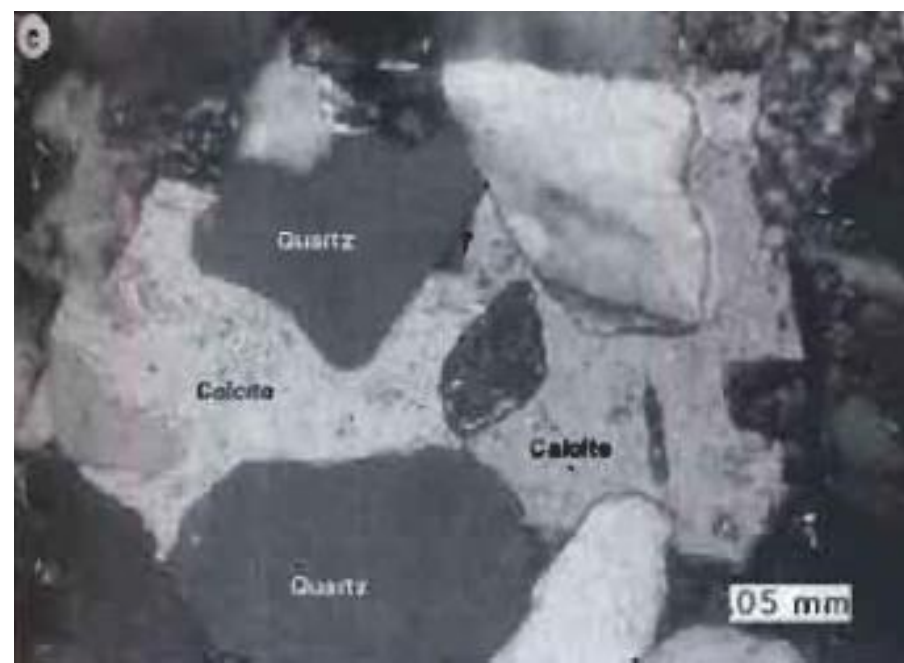
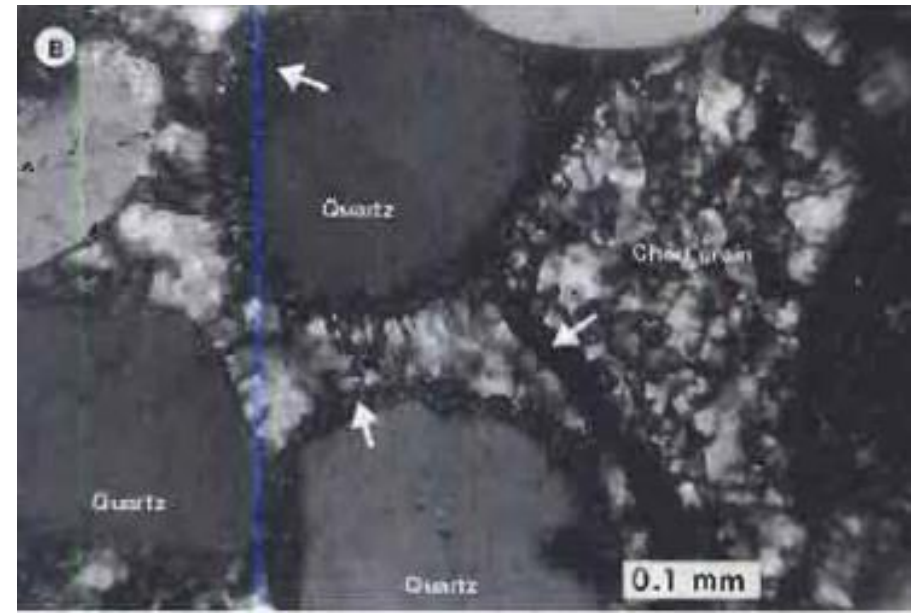
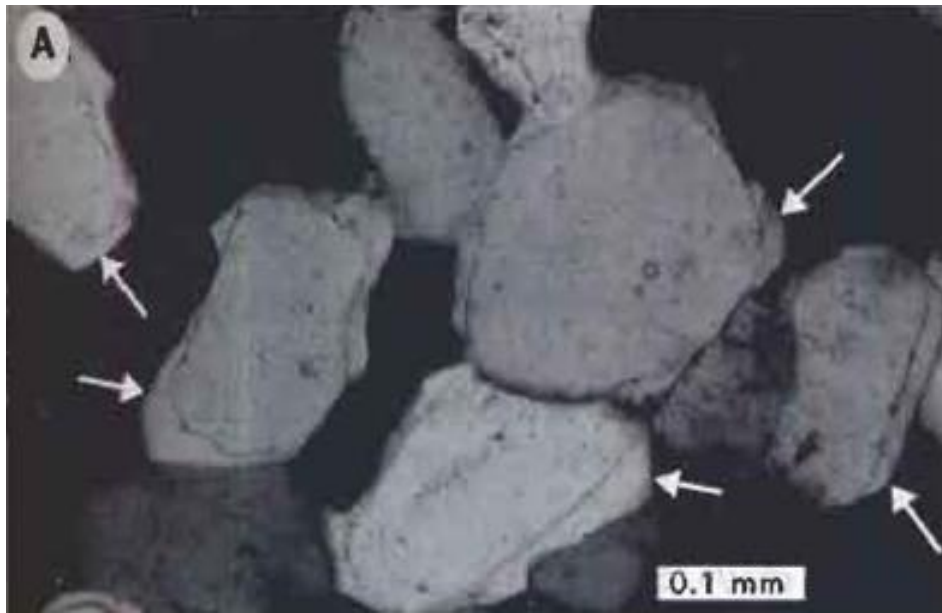
- New crystals form between grains
- Precipitation out of groundwater
- Fills in remaining porosity



From: northstonematerials.com

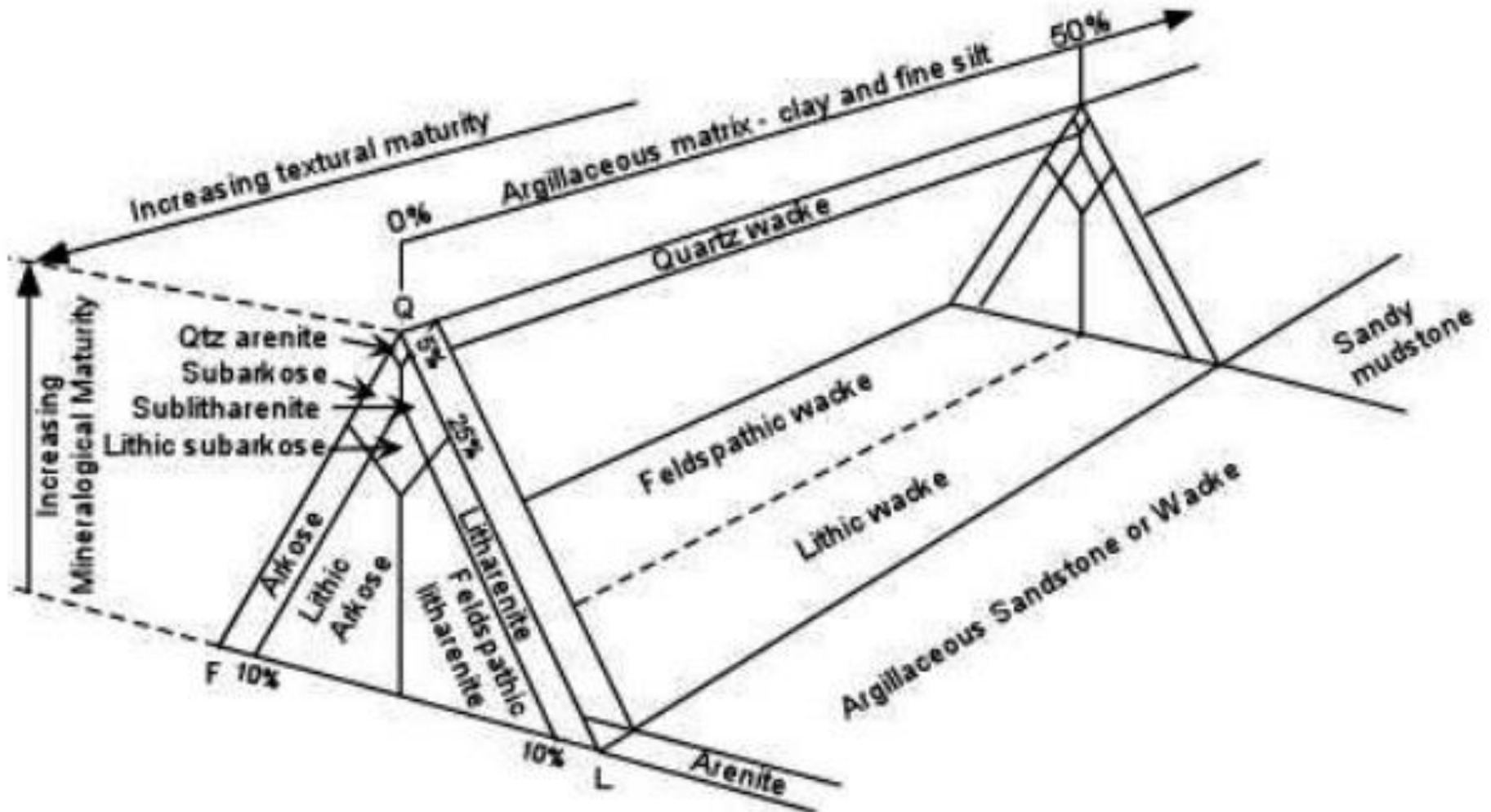


From: unibe.ch



Cement

Dott Classification



Conglomerate

Sediment $\xrightarrow{\text{Lithification}}$ Sedimentary rock





**Clast-Supported
Conglomerate**



Breccia

**Matrix-Supported
Conglomerate**



**Oligomictic
Conglomerate**



**Polymictic
Conglomerate**

